Roll No.

Total Pages : 3

239401

May, 2019

M.Sc. Chemistry, Semester-IV Inorganic Chemistry Special-I (CH-401-A)

Time : 3 Hours]

[Max. Marks: 75

Note :

- 1. It is compulsory to answer the questions of Part-1. Limit your answers within 20-40 words in this part.
- 2. Answer any four questions from Part-2 in detail.
- 3. Different parts of the same question are to be attempted adjacent to each other.
- Assume suitable standard data wherever required, if not given.

PART-1

- **1.** (a) What are anation reactions?
 - (b) What is principle of NAA?
 - (c) Define Quantum Yield.
 - (d) Explain Compound Nucleus Theory.
 - (e) Define isopoly anions with example.
 - (f) What are Redistribution reactions?

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(g) What are thermonuclear reactions?

(h) What do you mean by nuclear stability?

(i) What is carbonyl scrambling?

(j) What are Photochemical Laws?

 $(10 \times 1.5 = 15)$

PART-2

- 2. (a) Explain the inner sphere mechanism with example. 7.5
 - (b) Explain metal ion catalysed reactions with example. 7.5
- 3. (a) Explain the fluxional process in η^2 -allyl metal complex. 7.5
 - (b) What is Franck Condon Principle? Highlight its significance in Photochemistry. 7.5
- 4. Write short notes on the following :
 - (i) Photoisomerisation reactions. 7.5(ii) Photoredox processes conditions of excited state for
 - redox reactant. 7.5
- Describe the principle and application of Isotopic Dilution Analysis?
 15

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6. Explain various types of Nuclear Reactions. 15

- (a) State Grotthus Draper Law and give its significance in photochemistry. 7.5
 - (b) What is base hydrolysis? Explain conjugate base mechanism in detail. 7.5

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