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239202

May 2019

M.Sc. (Chemistry) IInd Semester ORGANIC CHEMISTRY (CH-202A)

Time: 3 Hours]

[Max. Marks: 75

Instructions:

- (i) It is compulsory to answer all the questions (1.5 marks each) of Part-A in short.
- (ii) Answer any four questions from Part-B in detail.
- (iii) Different sub-parts of a question are to be attempted adjacent to each other.

PART-A

- 1. (a) Give two evidences in the favour of cyclic bromonium ion formation as reaction intermediate in addition reactions of halogens to alkenes. (1.5)
 - (b) Draw the major product in the following reaction:

(1.5)

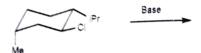
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- (c) Why p-nitroaniline cannot be obtained directly from aniline. (1.5)
- (d) Giving hybridization state of carbon free radicals, discuss their structure. (1.5)
- (e) Discuss briefly the factors affecting E1 reactions.
 - (1.5)

(1.5)

f) Write down the product;



- (g) Discuss the stereochemistry of E₂ elimination reactions.
- (h) Explain why NaH cannot be used for reduction of carbonyl compounds while it is stronger hydride ion donor than NaBH₄ or LiAlH₄. (1.5)
- (i) Draw potential energy level diagram and explain sigma-complex formation in aromatic electrophilic substitution reaction. (1.5)
- (j) Briefly explain why *m*-Xylene undergoes nitration 100 times faster than *p*-Xylene? (1.5)

PART-B

2. (a) Explain E1CB and [E1CB]ip mechanism of elimination reactions. (10)

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(b) Explain Saytzeff and Hoffmann elimination products giving suitable examples. (5)

- 3. (a) Discuss the factors governing the reactivity profile of aliphatic electrophilic substitutions. (8)
 - (b) Explain SE2 and SEi mechanisms of aliphatic electrophilic substitution reaction. (7)
- 4. Giving suitable mechanism, explain the following name reactions:
 - (a) Bischler Napierlaski Reaction. (5)
 - (c) Fries Rearrangement. (5)
 - (d) Vielsmeier Haawk Reaction. (5)
- 5. Complete the following reactions with plausible mechanism:

(e)
$$\frac{1}{1}$$
 OH $\frac{1}{1}$ (3x5)

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- 6. (a) Discuss the acid catalyzed acetal/hemiacetal formation. (5)
 - (b) Giving suitable mechanism, explain Reformatsky reaction. (5)
 - (c) Discuss the reactivity of hydrohalogen acids toward addition to C—C double bonds. (5)
- 7. (a) Write down the product of following reaction with mechanism: