

Roll No.

Total Pages : 3

751103

January 2023

M.Sc. (Chemistry) Ist SEMESTER

Physical Chemistry (General-I) (CH-103B)

Time : 3 Hours]

[Max. Marks : 75

Instructions :

- 1. It is compulsory to answer all the questions (1.5 marks each) of Part-A in short.*
- 2. Answer any four questions from Part-B in detail.*
- 3. Different sub-parts of a question are to be attempted adjacent to each other.*
- 4. Use of non programmable scientific calculator is allowed.*

PART-A

1. (a) Derive expression for $\left(\frac{d}{dx} + x\right)^2$. (1.5)
- (b) Write expression for thermodynamic probability for FD statistics. (1.5)
- (c) What will happen if walls of one dimensional box are suddenly removed? (1.5)
- (d) Write expectation values of x , x^2 , P_x , P_x^2 . (1.5)
- (e) Distinguish reversible and irreversible processes in thermodynamics. (1.5)

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- (f) Calculate minimum energy for an electron confined to a molecule of length 1 nm. (1.5)
- (g) Write Schrödinger equation for 3D box. (1.5)
- (h) What do you mean by eigen function and eigen values? (1.5)
- (i) Discuss criteria for spontaneity of a process in brief. (1.5)
- (j) Calculate the probability of finding the particle between 0.49a and 0.51a for Ψ_1 and Ψ_2 . (1.5)

PART-B

2. (a) Solve Schrödinger wave equation for hydrogen-like particles. (10)
- (b) Explain the concept of degeneracy in case of three dimensional box. (5)
3. (a) Explain quantum tunneling in details. (5)
- (b) Discuss free energy functions and their significance in details. (10)
4. Find expressions for most probable distribution of N number of particles among various energy levels according to BE statistics. (15)
5. (a) Do a comparison for M-B, BE and FD statistics. (5)
- (b) Discuss Schrödinger equation for a particle in a box and find expression for zero point energy, and normalized wave function and probability function. (10)

6. (a) Explain shapes of s, p and d-orbitals with the help of θ and Φ functions. (10)
- (b) Calculate value of Lagrange's undetermined multipliers α and β . (5)
7. What do you mean by partial molar quantities? Drive expression for Chemical Potential, free energy, and volume. Also derive Gibb's-Duhem equation. (15)
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